Acta Crystallographica Section E

## Structure Reports Online

ISSN 1600-5368

# *N,N'*-Di-2-pyridylperylene-3,4:9,10-bis(dicarboximide)

### Jin Mizuguchi,\* Kazuyuki Hino, Kazuyuki Sato, Hiroo Takahashi and Shigeru Suzuki

Department of Applied Physics, Graduate School of Engineering, Yokohama National University, 79-5 Tokiwadai, Hodogaya-ku, 240-8501 Yokohama, Japan

Correspondence e-mail: mizu-j@ynu.ac.jp

#### **Key indicators**

Single-crystal X-ray study  $T=93~\mathrm{K}$  Mean  $\sigma(\mathrm{C-C})=0.006~\mathrm{\mathring{A}}$  R factor = 0.062 wR factor = 0.045 Data-to-parameter ratio = 11.1

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

The title compound,  $C_{34}H_{16}N_4O_4$ , is a perylene–imide pigment utilized for  $H_2$  gas sensors. There are two independent half-molecules in the asymmetric unit. The symmetry of both molecules is  $C_i$ . The angles between the each of the pyridyl rings and the perylene-imide skeleton are 77.7 (1) and 72.8 (1) in the two molecules. The independent molecules are stacked alternately along the b axis.

Received 14 January 2005 Accepted 19 January 2005 Online 29 January 2005

#### Comment

Perylene compounds are industrially important pigments, covering a variety of shades from red *via* maroon to black (Herbst & Hunger, 1993). The title compound, (I), is an *ortho*pyridyl derivative, abbreviated to OPP.

$$R = N$$
 $R = N$ 
 $R =$ 

We have carried out a series of investigations on H<sub>2</sub> gas sensors utilizing a high proton affinity of organic pigments that have pyridyl rings connected directly to the choromophore (Takahashi & Mizuguchi, 2005). The N atom of the pyridyl ring works as a strong proton acceptor and plays a decisive role for signal detection. OPP exhibits a resistivity change of about three orders of magnitude even for 0.05% H<sub>2</sub>. Other than the present *ortho* derivative, there are also *meta* and *para* derivatives and their sensitivity is slightly different, depending the site of the N atom. In this connection, structure analyses of these derivatives have been carried out. The structures of the *meta* and *para* derivatives are reported in the preceding and following papers, respectively (Mizuguchi *et al.*, 2005; Hino *et al.*, 2005). This paper reports the structure of the *ortho* derivative.

The two independent molecules, A and B, of OPP (Fig. 1) are characterized by the same molecular  $C_i$  symmetry. The molecular conformations of these molecules are quite similar, but the twist angle of the pyridyl rings is different. The angles between each of the pyridyl rings and the perylene-imide

© 2005 International Union of Crystallography Printed in Great Britain – all rights reserved

Figure 1 C17
A view of the molecular conformations of the two independent molecules of (I), showing 50% displacement ellipsoids for the non-H atoms. Unlabelled atoms in A are related to labelled atoms by -x, 1-y, 1-z; unlabelled atoms in B are related to labelled atoms by 2-x, 1-y, -z.

O1-C1

skeleton are 77.7 (1)° in molecule A and 72.8 (1)° in molecule B. The perylene–imide skeleton is planar (r.m.s. deviations = 0.048 and 0.032 Å for the two molecules in the asymmetric unit). Molecules A and B are stacked alternately along the b axis, as shown in Fig. 2.

#### **Experimental**

OPP was synthesized by reaction of perylenetetracarboxylic dianhydride with 1,2-di-2-pyridyldiamine in dimethylnaphthalene at 490 K for 3 h according to the method of Herbst & Hunger (1993). The product was then purified three times by sublimation at 760 K, using a two-zone furnace (Mizuguchi, 1981). Single crystals of OPP were grown from the vapor phase in a closed system based on a two-zone furnace. After 48 h, a number of single crystals were obtained in the form of needles.

Crystal	data

-	
$C_{34}H_{16}N_4O_4$	$D_x = 1.485 \text{ Mg m}^{-3}$
$M_r = 544.51$	Cu Kα radiation
Monoclinic, $P2_1/c$	Cell parameters from 12013
a = 17.599 (1)  Å	reflections
b = 7.1705 (5)  Å	$\theta = 3.6-65.6^{\circ}$
c = 20.679 (2)  Å	$\mu = 0.82 \text{ mm}^{-1}$
$\beta = 111.004 (5)^{\circ}$	T = 93.2  K
$\beta = 111.004 (5)^{\circ}$ $V = 2436.2 (3) \text{ Å}^3$	Needle, red
Z = 4	$0.50 \times 0.09 \times 0.05 \text{ mm}$
Data collection	
Rigaku R-AXIS RAPID Imaging	4219 independent reflections
Plate diffractometer	2120 reflections with $F^2 > 2\sigma(F^2)$
$\omega$ scans	$R_{\rm int} = 0.051$
Absorption correction: multi-scan	$\theta_{\rm max} = 68.2^{\circ}$
/ . <del>-</del>	

#### Refinement

(ABSCOR; Higashi, 1995)

 $T_{\min} = 0.960, \ T_{\max} = 0.960$ 

20469 measured reflections

Refinement on  $F^2$  H-atom parameters constrained  $R[F^2>2\sigma(F^2)]=0.062$   $w=1/[\sigma^2(F_o^2)]$   $wR(F^2)=0.045$   $(\Delta/\sigma)_{\rm max}=0.001$  S=1.89  $\Delta\rho_{\rm max}=0.47~{\rm e}~{\rm \mathring{A}}^{-3}$  4219 reflections  $\Delta\rho_{\rm min}=-0.49~{\rm e}~{\rm \mathring{A}}^{-3}$  379 parameters

 $h=-21\to 21$ 

 $k = -7 \rightarrow 7$ 

 $l = -24 \rightarrow 24$ 

**Table 1** Selected geometric parameters (Å, °).

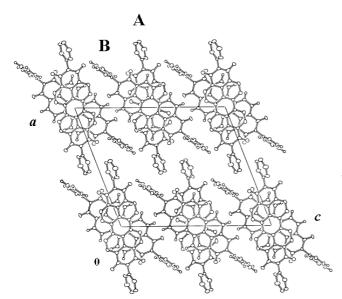
1.217 (4)

C8-C9

01-01	1.217 (4)	C8-C9	1.392 (3)	
O2-C11	1.202 (4)	C9-C10	1.378 (5)	
O3-C18	1.196 (4)	C10-C11	1.494 (5)	
O4-C28	1.217 (4)	C10-C12	1.414 (4)	
N1-C1	1.422 (4)	C18-C19	1.506 (5)	
N1-C11	1.418 (4)	C19-C20	1.377 (5)	
N3-C18	1.399 (4)	C19-C29	1.405 (4)	
N3-C28	1.434 (4)	C20-C21	1.401 (5)	
C1-C2	1.496 (5)	C21-C22	1.390 (4)	
C2-C3	1.382 (4)	C22-C23	1.435 (5)	
C2-C12	1.417 (4)	C22-C24 <sup>ii</sup>	1.480 (5)	
C3-C4	1.385 (5)	C23-C24	1.427 (4)	
C4-C5	1.401 (5)	C23 – C29	1.437 (5)	
C5-C6	1.447 (4)	C24—C25	1.393 (5)	
C5-C7 <sup>i</sup>	1.468 (5)	C25—C26	1.393 (5)	
C6-C7	1.437 (5)	C25—C20 C26—C27	1.395 (4)	
C6-C12	` '			
	1.427 (5)	C27—C28	1.487 (5)	
C7—C8	1.394 (4)	C27—C29	1.420 (5)	
C1-N1-C11	127.6 (4)	C6-C12-C10	119.4 (4)	
C18-N3-C28	125.5 (4)	O3-C18-N3	120.9 (4)	
O1-C1-N1	121.5 (4)	O3-C18-C19	123.5 (4)	
O1-C1-C2	123.8 (4)	N3-C18-C19	115.5 (4)	
N1-C1-C2	114.8 (4)	C18-C19-C20	117.8 (4)	
C1-C2-C3	118.3 (4)	C18-C19-C29	121.3 (4)	
C1-C2-C12	120.5 (4)	C20-C19-C29	120.9 (4)	
C3-C2-C12	121.1 (4)	C19-C20-C21	120.1 (4)	
C2-C3-C4	120.6 (4)	C20-C21-C22	121.5 (4)	
C3-C4-C5	121.0 (4)	C21 – C22 – C23	119.4 (4)	
C4-C5-C6	119.5 (4)	C21 - C22 - C23 C21 - C22 - C24 ii	122.2 (4)	
C4-C5-C7 <sup>i</sup>	121.9 (4)	C23-C22-C24 <sup>ii</sup>	118.4 (4)	
C6-C5-C7 <sup>i</sup>	118.6 (4)	C22-C23-C24	122.9 (4)	
C5-C6-C7	122.8 (4)	C22-C23-C24 C22-C23-C29	118.3 (4)	
C5-C6-C12	118.5 (4)	C24—C23—C29		
C7-C6-C12	` '	C24—C25—C29 C22 <sup>ii</sup> —C24—C23	118.8 (4)	
C7 - C6 - C12 $C5^{i} - C7 - C6$	118.6 (4)	C22 - C24 - C25 C22 <sup>ii</sup> - C24 - C25	118.7 (4)	
C5 - C7 - C6 $C5^{i} - C7 - C8$	118.6 (4)		121.4 (4)	
	121.7 (4)	C23-C24-C25	119.9 (4)	
C6-C7-C8	119.8 (4)	C24—C25—C26	121.3 (4)	
C7-C8-C9	120.7 (4)	C25-C26-C27	120.2 (4)	
C8-C9-C10	120.9 (4)	C26-C27-C28	119.5 (4)	
C9-C10-C11	118.4 (4)	C26-C27-C29	120.4 (4)	
C9-C10-C12	120.5 (4)	C28-C27-C29	120.1 (4)	
C11-C10-C12	121.1 (4)	O4-C28-N3	120.4 (4)	
O2-C11-N1	120.7 (4)	O4-C28-C27	123.3 (4)	
O2-C11-C10	125.0 (4)	N3-C28-C27	116.3 (4)	
N1-C11-C10	114.3 (4)	C19-C29-C23	119.8 (4)	
C2-C12-C6	119.2 (4)	C19-C29-C27	120.8 (4)	
C2-C12-C10	121.3 (4)	C23-C29-C27	119.3 (4)	
Symmetry codes: (i) $-x + 1 - y + 1 - z$ ; (ii) $2 - x + 1 - y - z$				

Symmetry codes: (i) -x, 1 - y, 1 - z; (ii) 2 - x, 1 - y, -z.

1.392 (5)



**Figure 2** Projection of the structure on to the *ac* plane.

All H atoms were positioned geometrically [C-H = 0.95 Å and  $U_{\rm iso}$  = 1.2 $U_{\rm eq}$ (C)] and refined using a riding model.

Data collection: *PROCESS-AUTO* (Rigaku, 1998); cell refinement: *PROCESS-AUTO*; data reduction: *TEXSAN* (Molecular Structure Corporation, 2001); program(s) used to solve structure: *SHELXS86* (Sheldrick, 1985); program(s) used to refine structure: *TEXSAN*; molecular graphics: *ORTEPIII* (Burnett & Johnson, 1996); software used to prepare material for publication: *TEXSAN*.

The authors are indebted to Mr I. Suzuki for experimental assistance.

#### References

Burnett, M. N. & Johnson, C. K.(1996). *ORTEPIII*. Report ORNL-6895. Oak Ridge National Laboratory. Tennessee, USA.

Herbst, W. & Hunger, K. (1993). *Industrial Organic Pigments*, pp. 467–475. Weinheim: VCH.

Higashi, T.(1995). ABSCOR. Rigaku Corporation, Tokyo, Japan.

Hino, K., Sato, K., Takahashi, H., Suzuki, S. & Mizuguchi, J. (2005). *Acta Cryst.* E**61**, 0440–0441.

Mizuguchi, J. (1981). Cryst. Res. Technol. 16, 695-700.

Mizuguchi, J., Hino, K., Sato, K., Takahashi, H. & Suzuki, S. (2005). *Acta Cryst.* E**61**, o434–o436.

Molecular Structure Corporation (2001). *TEXSAN*. Version 1.11. MSC, 9009 New Trails Drive, The Woodlands, TX 77381-5209, USA.

Rigaku (1998). PROCESS-AUTO. Rigaku Corporation, Tokyo, Japan.

Sheldrick, G. M. (1985). SHELXS86. Crystallographic Computing 3, edited by G. M. Sheldrick, C. Krüger and R. Goddard, pp. 175–189. Oxford University Press.

Takahashi, H. & Mizuguchi, J. (2005). J. Electrochem. Soc. In the press.